

Tin-mediated enantioselective aldol-type reaction for the asymmetric synthesis of α -substituted serines utilizing an external chiral ligand, (-)-sparteine

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Abstract

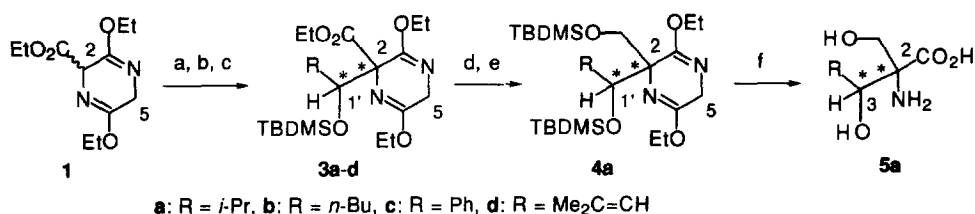
The enantioselective aldol-type reaction of ethyl 3,6-diethoxy-2,5-dihydro-2-pyrazinecarboxylate (**1**) with achiral aldehydes **2a-d** was investigated by employing $\text{Sn}(\text{OSO}_2\text{CF}_3)_2$ -triethylamine in the presence of (-)-sparteine. Not only a stoichiometric amount, but also 0.3 mol equiv. of (-)-sparteine promoted the highly enantioselective aldol-type reaction. © 1999 Elsevier Science B.V. All rights reserved.

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The asymmetric synthesis of α -substituted α -amino acids is a topic of current interest,¹⁻³ the synthesis of α -substituted serines in particular being a major focus of study in recent years. We have previously reported diastereoselective reactions and chemoenzymatic enantiodivergent reactions useful for the asymmetric synthesis of α -substituted serines.⁴⁻⁸ In these studies, the chiral recognition mode with $\text{Sn}(\text{OSO}_2\text{CF}_3)_2$ -*N*-ethylpiperidine⁹ was demonstrated to be quite different from that with MgBr_2 -triethylamine in the diastereoselective aldol-type reaction of (*5R* or *5S*)-3,6-diethoxy-2,5-dihydro-5-isopropyl-2-pyrazinecarboxylate with various aldehydes.^{5,8} We now describe the Sn(II)-mediated enantioselective aldol-type reaction of ethyl 3,6-diethoxy-2,5-dihydro-2-pyrazinecarboxylate (**1**) employing (-)-sparteine¹⁰ as an external chiral ligand (Scheme 1). The bislactim ether **1** was readily prepared from σ -symmetric diethyl aminomalonate and glycine.⁵

First of all, the diastereoselective mode of the Sn(II)- or Mg(II)-mediated aldol-type reaction of the newly designed achiral bislactim ether **1** with achiral aldehydes **2a-d**

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Scheme 1 a) Sn(II) or Mg(II): see footnote for Table 1-4, b) RCHO (**2a-d**), c) Me₂(*t*-Bu)SiOSO₂CF₃ / 2,6-lutidine / CH₂Cl₂ / -45 °C, d) DIBAL / toluene / 0 °C, e) Me₂(*t*-Bu)SiOSO₂CF₃ / 2,6-lutidine / CH₂Cl₂ / rt, f) 12N HCl - CF₃CO₂H - *i*-PrOH (2 : 1 : 1) / 150 °C.

Table 1
Diastereoselective aldol-type reaction of bislactim ether **1** with aldehydes **2a-d**

Entry	Aldehyde	Conditions ^{a)}	Time (h)	Yield (%) ^{b)} of 3a-d	Diastereomer Ratio ^{c)} (2 <i>S</i> [*] ,1' <i>R</i> [*]) : (2 <i>S</i> [*] ,1' <i>S</i> [*])
1	2a	A	2	74	84 : 16
2	2a	B	3.5	82	5 : 95
3	2b	A	2	87	84 ^{d)} : 16 ^{d)}
4	2b	B	2.5	91	5 ^{d)} : 95 ^{d)}
5	2c	A	3.5	73	4 ^{d)} : 96 ^{d)}
6	2c	B	2.5	80	21 ^{d)} : 79 ^{d)}
7	2d	A	4	85	47 ^{d)} : 53 ^{d)}
8	2d	B	2.5	77	7 ^{d)} : 93 ^{d)}

a) A: THF, -78 °C, **1** / Sn(OSO₂CF₃)₂ / *N*-ethylpiperidine / **2** (1 : 1.5 : 1.6 : 2); B: MeCN, -20 °C, **1** / MgBr₂ / triethylamine / **2** (1 : 2 : 4 : 2). b) Total isolation yield of diastereomers. c) ¹H NMR analysis (400 MHz NMR, CDCl₃). d) Relative configuration was assigned arbitrarily.

Table 2
Enantioselective aldol-type reaction of bislactim ether **1** with aldehydes **2a-d** in the presence of (-)-sparteine (1.5 mol equiv.) ^{a)}

Entry	Aldehyde	Time (h)	Yield (%) ^{b)} of 3a-d	Diastereomer Ratio ^{c)} (2 <i>S</i> [*] ,1' <i>R</i> [*]) : (2 <i>S</i> [*] ,1' <i>S</i> [*])	Ee (%) ^{d)} of (2 <i>S</i> [*] ,1' <i>R</i> [*])- 3a-d
1 ^{e)}	2a	2	90	88 : 12	98
2	2a	2	89	82 : 18	95
3	2b	2	83	52 ^{f)} : 48 ^{f,g)}	98
4	2c	3.5	72	93 ^{f)} : 7 ^{f)}	95
5	2d	4	79	71 ^{f)} : 29 ^{f)}	84

a) CH₂Cl₂, -78 °C, **1** / Sn(OSO₂CF₃)₂ / triethylamine / (-)-sparteine / **2** (1 : 1.5 : 1.6 : 1.5 : 2). b) Total isolation yield of diastereomers. c) ¹H NMR analysis (400 MHz NMR, CDCl₃). d) HPLC analysis (Chiralcel OD, hexane / 2-propanol). e) *N*-ethylpiperidine was employed instead of triethylamine. f) Relative configuration was assigned arbitrarily. g) (2*S*^{*},1'*S*^{*})-**3b**: 84% ee.

proved to be similar to that with (*5R* or *5S*)-3,6-diethoxy-2,5-dihydro-5-isopropyl-2-pyrazinecarboxylate (Table 1).^{5,8} Namely, the aldol-type reaction with aliphatic aldehydes **2a,b** employing Sn(OSO₂CF₃)₂-*N*-ethylpiperidine or MgBr₂-triethylamine gave the different diastereomers **3a,b** as each major product after silylation of the aldol products (Table 1, entries 1-4). On the other hand, each major product was the same diastereomers **3c,d** in the similar Sn(II)- or Mg(II)-mediated reaction with unsaturated aldehydes **2c,d** (Table 1, entries 5-8). The diastereomer ratios of **3a-d** were determined by employing ¹H NMR analysis (400 MHz, CDCl₃).

Based on these results, the Sn(II)-mediated enantioselective aldol-type reactions of bislactim ether **1** with achiral aldehydes **2a-d** in the presence of a stoichiometric amount of (-)-sparteine were examined, as shown in Table 2. The aldol-type reactions of **1** with achiral aldehydes **2a-d** (2 mol equiv.) using Sn(OSO₂CF₃)₂ (1.5 mol equiv.) and triethylamine (1.6 mol equiv.) in the presence of (-)-sparteine (1.5 mol equiv.) in CH₂Cl₂ at -78 °C afforded compounds **3a-d** with enantioselectivities of up to 98% ee. The ee values of **3a-d** were determined by HPLC analysis (Daicel Chiralcel OD, hexane / 2-propanol). The absolute configuration of the major aldol product **3a** was confirmed to be (*2S,1'R*)-**3a** by its chemical conversion to the known α -substituted serine, as shown in Scheme 1.⁸ Namely, reduction (36%) of **3a** (98% ee) with DIBAL at 0 °C followed by protection (59%) of the hydroxyl group with TBDMSOTf gave a diastereomerically pure **4a**, which was submitted to hydrolysis (46%) to obtain (*2S,3R*)-**5a** {[α]_D²⁴ +29.7 (*c* 0.13, MeOH), lit.⁸ (*2R,3S*)-**5a** [α]_D²⁷ -26.1 (*c* 0.12, MeOH)}. The excellent Sn(II)-mediated asymmetric induction with **2a,b** (Table 2, entries 1 and 3) prompted us to examine the use of a catalytic amount of (-)-sparteine in these aldol-type reactions.

Table 3
Enantioselective aldol-type reaction of bislactim ether **1** with aldehyde **2a** in the presence of (-)-sparteine (0.3 mol equiv.)^{a)}

Entry	Triethylamine (mol equiv.)	Yield (%) ^{b)} of 3a	Diastereomer Ratio ^{c)} (<i>2S',1'R'</i>) : (<i>2S'',1'S''</i>)	Ee (%) ^{d)} of (<i>2S,1'R</i>)- 3a
1	1.6	65	80 : 20	68
2	4	64	80 : 20	76
3	6	65	84 : 16	80
4	8	69	85 : 15	94
5	16	62	84 : 16	93

a) CH₂Cl₂, -78 °C, 2 h, **1** / Sn(OSO₂CF₃)₂ / (-)-sparteine / **2** (1 : 1.5 : 0.3 : 2). b) Total isolation yield of diastereomers. c) ¹H NMR analysis (400 MHz NMR, CDCl₃). d) HPLC analysis (Chiralcel OD, hexane / 2-propanol).

The results summarized in Table 3 revealed that 0.3 mol equiv. of (-)-sparteine induced high enantioselectivity in the presence of excess triethylamine (Table 3, entries 4 and 5). The scope of the reaction employing 0.3 mol equiv. of (-)-sparteine in the presence of 8 mol equiv. of triethylamine was also investigated, as shown in Table 4. The aldol-type

reactions with aldehydes **2a,b** afforded the corresponding chiral compounds **3a,b** in 94 and 92% ee (Table 4, entries 2 and 3), respectively. Although excellent stereocontrol can be achieved in the Sn(II)-mediated aldol-type reaction of enolsilanes,¹¹⁻¹³ to our knowledge, our method is the first example of an enantioselective aldol-type reaction involving the *in situ* formation of Sn(II)-enolate employing a catalytic amount of an external chiral ligand. Enolsilanes are not necessary to perform these catalytic enantioselective aldol-type reactions. We are currently pursuing studies on the reaction mechanism as well as on the design of new external chiral ligands.

Table 4
Enantioselective aldol-type reaction of bislactim ether **1** with aldehydes **2a-d** in the presence of (-)-sparteine (0.3 mol equiv.)^{a)}

Entry	Aldehyde	Time (h)	Yield (%) ^{b)} of 3a-d	Diastereomer Ratio ^{c)} (2 <i>S</i> [*] ,1' <i>R</i> [*]) : (2 <i>S</i> [*] ,1' <i>S</i> [*])	Ee (%) ^{d)} of (2 <i>S</i> ,1' <i>R</i>)- 3a-d
1 ^{e)}	2a	2	59	85 : 15	88
2	2a	2	69	85 : 15	94
3	2b	2	65	43 ^{f)} : 57 ^{f,g)}	92
4	2c	3.5	57	91 ^{f)} : 9 ^{f)}	78
5	2d	4	67	74 ^{f)} : 26 ^{f)}	66

a) CH₂Cl₂, -78 °C, **1** / Sn(OSO₂CF₃)₂ / triethylamine / (-)-sparteine / **2** (1 : 1.5 : 8 : 0.3 : 2). b) Total isolation yield of diastereomers. c) ¹H NMR analysis (400 MHz NMR, CDCl₃). d) HPLC analysis (Chiralcel OD, hexane / 2-propanol). e) *N*-ethylpiperidine was employed instead of triethylamine. f) Relative configuration was assigned arbitrarily. g) (2*S*^{*},1'*S*^{*})-**3b**: 68% ee.

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